Kai-Chi Chang et. al

Supplementary Data

Highly selective fluorescence sensing of Cu²⁺ ion by an arylisoxazole modified

calix[4]arene

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Table of contents

Page No

Content

- S2. ¹H and ¹³C NMR Spectra of $\mathbf{1}$.
- S3. ¹H and ¹³C NMR Spectra of $\mathbf{2}$.
- S4. ¹H and ¹³C NMR Spectra of 4.
- S5. UV/vis spectra of **1** and **2** (20 μ M) in CH₃CN/CHCl₃ (v/v = 1000:4) (Figure S7) & Changes in fluorescence emission spectra of **2** (20 μ M) before and after adding 200 μ M concentration of various metal perchlorates in CH₃CN/CHCl₃ (v/v = 1000:4) (Figure S8).
- S6 UV/vis spectra of **1** (20 μ M) before and after adding a 200 μ M concentration of Cu(ClO₄)₂ in CH₃CN/CHCl₃ (v/v = 1000:4) (Figure S9) & Fluorescence emission spectra of **1** (20 μ M) upon addition of various equiv of Cu(ClO₄)₂ in DMSO (Figure S10).
- S7. Fluorescence emission spectra of **1** (20 μ M) upon addition of various equiv of Cu(ClO₄)₂ in MeOH/CHCl₃ (v/v = 1000:4). (Figure S11) & Stern-Volmer plot of **1** with Cu(ClO₄)₂ (Figure S12).
- S8. ¹H NMR spectra of **2** (5.0 mM) in CDCl₃/CD₃CN (3/1) (a) and in the presence of 25 mM (5.0 equiv) of Cu(ClO₄)₂ (b), where * denotes an external standard CHCl₃ (Figure S13) & Changes in fluorescence emission spectra of **1** (20 μ M) before and after adding 200 μ M concentration of [(CH₃CN)₄Cu]PF₆ (dash curve) and Cu(ClO₄)₂ (short-dash curve) in CH₃CN/CHCl₃ (v/v = 1000:4) (Figure S14).
- S9. Fluorescence emission changes for the **1** (20 μ M) with 3 equiv of Cu(ClO₄)₂ in CH₃CN/CHCl₃ (v/v = 1000:4) upon addition of various amounts of [(CH₃CN)₄Cu]PF₆ (Figure S15).



Figure S1. ¹H NMR (300 MHz) Spectrum of 1 in CDCl₃.



Figure S2. ¹³C NMR (75.4 MHz) spectrum of 1 in CDCl₃.



Figure S3. ¹H NMR (300 MHz) spectrum of 2 in CDCl₃.



Figure S4. ¹³C NMR (75.4 MHz) spectrum of 2 in CDCl₃..



Figure S5. ¹H NMR (300 MHz) spectrum of 4 in CDCl₃.



Figure S6. 13 C NMR (75.4 MHz) spectrum of **4** in CDCl₃.



Figure S7. UV/vis spectra of 1 and 2 (20 μ M) in CH₃CN/CHCl₃ (v/v = 1000:4).



Figure S8. Changes in fluorescence emission spectra of **2** (20 μ M) before and after adding 200 μ M concentration of various metal perchlorates in CH₃CN/CHCl₃ (v/v = 1000:4).



Figure S9. UV/vis spectra of **1** (20 μ M) before and after adding a 200 μ M concentration of Cu(ClO₄)₂ in CH₃CN/CHCl₃ (v/v = 1000:4).



Figure S10. Fluorescence emission spectra of 1 (20 μ M) upon addition of various equiv of Cu(ClO₄)₂ in DMSO.



Figure S11. Fluorescence emission spectra of **1** (20 μ M) upon addition of various equiv of Cu(ClO₄)₂ in MeOH/CHCl₃ (v/v = 1000:4).



Figure S12. Stern-Volmer plot of 1 with Cu(ClO₄)₂.



Figure 13. ¹H NMR spectra of **2** (5.0 mM) in CDCl₃/CD₃CN (3/1) (a) and in the presence of 25 mM (5.0 equiv) of Cu(ClO₄)₂ (b), where * denotes an external standard CHCl₃.



Figure S14. Changes in fluorescence emission spectra of **1** (20 μ M) before and after adding 200 μ M concentration of [(CH₃CN)₄Cu]PF₆ (dash blue curve) and Cu(ClO₄)₂ (short-dash curve) in CH₃CN/CHCl₃ (v/v = 1000:4).



Figure S15. Fluorescence emission changes for the **1** (20 μ M) with 3 equiv of Cu(ClO₄)₂ in CH₃CN/CHCl₃ (v/v = 1000:4) upon addition of various amounts of [(CH₃CN)₄Cu]PF₆.